



# FOOD CHEMISTRY

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## Physico-chemical changes during baking

BINITA RANI  
ASSOCIATE PROFESSOR (DAIRY CHEMISTRY)  
FACULTY OF DAIRY TECHNOLOGY  
S.G.I.D.T., BVC CAMPUS,  
P.O.- BVC, DIST.-PATNA-800014



Several stages can be distinguished during the changes from **dough** to a **baked product**. They are :

- ❖ **Enzyme active stage (from 30<sup>0</sup>C to 70<sup>0</sup>C)**
- ❖ **Stage of starch gelatinization (From 55 to 70<sup>0</sup>C)**
- ❖ **Stage of water evaporation**
- ❖ **Stage of browning and aroma formation**

- ❖ These changes are different in  $\Rightarrow$  outer portion of the dough and in the interior of the crumb.
- ❖ This  $\Rightarrow$  because in the oven since heat transfer occurs slowly in the dough  $\Rightarrow$  there is a steep temperature gradient inward from crust of the dough.

**sequences of changes** taking place during conversion of foamy texture of dough => spongy texture of bread and other product by baking at a temperature of  $220^{\circ}\text{C}$  to  $250^{\circ}\text{C}$  include

### **Chemical and physical changes:**

#### **1. When dough is put in the oven**

- ❖ **rate of fermentation** initially **increases** as heat is conducted through the dough => Upto  $50^{\circ}\text{C}$ , yeast produces **CO<sub>2</sub>** and **ethanol**.
- ❖ **viscosity** of dough **falls** rapidly and reaches to **minimum** at about  $60^{\circ}\text{C}$ .
- ❖ At the same time **thermal expansion** of **gas** within each cell result in **rapid expansion of loaf volume**, known as “**oven spring**”.

2. As **internal temperature of dough increases above 37°C** => activity of yeast **decreases** and gets **inactivated** at 54°C.

❖ beyond 60°C, **viscosity** of dough **again increases rapidly** => caused by **swelling of starch** accompanied by **release of amylose** and by **protein denaturation**.

❖ As the **crumb starch gelatinizes** at 65°C => **α and β-amylase** present will **attack** => starch => amyolytic activity **continues** until their enzymes are **inactivated** at about 74 °C.

❖ optimum amyolytic activity is **desirable** to limit the degradation of gelatinized starch to counteract **staling of bread**.

- ❖ At the same time => denatured protein, swollen and partially gelatinized starch forms a **stable crumb network** at about 74°C => this transformation continues until the end of baking when the internal temp reaches to **93 - 100°C**.
- ❖ During this time **gluten** loses its tough and elastic state and becomes **stiff and brittle**.
- ❖ This **stiffens** starch structure so that a **firm elastic crumb** is formed => **starch granules** of crust surface **gelatinize** almost completely.
- ❖ This is specially the case when “oven humidity” is **high** => **resultant starch film** produces a **pleasing glaze**.
- ❖ This also **retards drying and settling** of the crust and permits **full expansion** of dough.

3. above process results in  $\Rightarrow$  **tremendous increase** in the **tensile strength** of dough and **increase**  $\Rightarrow$  presence of gas bubbles.

❖ Consequently **membrane** gives way and **becomes permeable**  $\Rightarrow$  allowing **H<sub>2</sub>O, CO<sub>2</sub> and ethanol** to **evaporate**  $\Rightarrow$  results  $\Rightarrow$  **baking weight loss**.

❖ **internal** temperature **never exceeds 100°C**, but the **outer** temperature reaches nearly **oven temperature (~ 200°C)**  $\Rightarrow$  **water evaporates** more from the surface and the **crust is formed**  $\Rightarrow$  results in **weight losses** during crust formation up to **8-14%** of the fresh dough weight.

- ❖ At high temperature to which the outer part of the dough is exposed => **Starch degrades** => dextrin, mono and disaccharide at 110<sup>0</sup>C-140<sup>0</sup>C.
- ❖ Caramelization and non-enzymatic browning also occur at ~140-150<sup>0</sup>C providing => sweetness and colours to the crust.
- ❖ roasted flavors develop at 150-200<sup>0</sup>C.
- ❖ In the crust => heterocyclic compounds pyrroline and pyridine, as well as furanone and 2 and 3 methyl butanal are formed => responsible for the roasty, malty and caramel flavor respectively in the products.
- ❖ autoxidation products of linoleic acid => such as methional, and diacetyl are also involved in the aroma of the crumb.



# Changes during storage

Bread quality rapidly changes during storage due to

- 1) **Moisture adsorption** - the crust loses its crispiness and glossiness.
- 2) **Aroma compounds of freshly baked bread evaporates** => resulting in **loss of flavor**.
- 3) Some of **very labile aroma compounds decrease rapidly** on storage due to oxidation and other reactions.
- 4) The **crumb structure also changes** => although at a slower rate.

- ❖ crumb becomes **firm** => its **elasticity** and **juiciness** => **lost**, and it **crumbles easily**.
- ❖ This is known as => **staling defect** of crumb, which is basically a **starch retrogradation phenomenon**

# Determination of gluten content in flour

- ❖ **gluten content** determine  $\Rightarrow$  **loaf volume** of bread and an increase in the **loaf volume** of bread is noticed  $\Rightarrow$  increase in **gluten content** of the flour.
- ❖ Gluten exhibits properties of  $\Rightarrow$  **cohesion, elasticity and viscosity** which are  $\Rightarrow$  combined characteristics of its two insoluble component proteins  $\Rightarrow$  **glutenin** and **gliadin**.
- ❖ For good **bread flour**:
- ❖ **wet gluten content** ranges between 30 to 36 % and
- ❖ **dry gluten content** ranges between 10 to 12 %.

## **Apparatus:**

1. Mortar
2. Glass rod
3. Hot air oven
4. Desiccator
5. Analytical balance

## Principle:

- ❖ **Gluten** is separated out from the flour by **washing the dough** made using water.
- ❖ albumins, globulins and other smaller proteins as well as **starch** are **washed away** with water leaving behind a **cohesive, elastic and rubbery** mass called => **crude wet gluten**.
- ❖ **65-75% of water** present in crude wet gluten is **dried** out by drying at **100 °C for 24 h** and weighed to get a value of **dry gluten**.

## Procedure:

25 g of flour + 15 ml of water => **mix** into a smooth and tight dough using fingers => **immerse** the dough ball into water for about 1 h to ensure proper hydration => **remove** the dough ball and **place** it on a piece of blotting silk cloth having an aperture of 0.5 mm => **wash** it with a gentle stream of water till the water passing through the silk cloth does not contain starch => **collect** the residue on the silk cloth and make it free of water by rubbing between dry palms or by using a suitable press = weigh as **wet gluten** => dry this wet gluten in an oven maintained at 100 °C for 24 h.

## Observations:

1. Weight of sample taken =  $W_1$  g
2. Weight of wet gluten =  $W_2$  g
3. Weight of dry gluten =  $W_3$  g

## Calculations:

- A. % wet gluten =  $(W_2/W_1) \times 100$
- B. % dry gluten =  $(W_3/W_1) \times 100$

# Determination of starch content in flour

- ❖ In wheat flour => **starch granules** are embedded in protein matrix.
- ❖ **major role** of starch => act as a **water sink** and set the system through **partial gelatinization**.
- ❖ Starch is also responsible for **staling phenomenon** since **amylose fraction retrogrades rapidly** during initial cooling of bread loaves. **Slow changes** in the **amylo-pectin fraction** are implicated in the further **firming** of bread **during storage**.
- ❖ Some of normal starches get **damaged** during milling stage => moderate amount of damaged starch is advisable while presence of **excessive damaged starch** => quite harmful



## **Apparatus:**

1. Conical Flasks
2. Funnel
3. Filter papers
4. Beakers

## **Reagents:**

1. Fehling A and B solutions
2. Methylene blue indicator
3. Concentrated HCl
4. Standard glucose solution
5. 50% NaOH
6. Phenolphthalein indicator

## Principle:

- ❖ flour  $\Rightarrow$  suspended in water and undissociated residue containing starch  $\Rightarrow$  allowed to hydrolyze in the presence of dilute HCl.
- ❖ glucose produced  $\Rightarrow$  filtered out and titrated against Fehling A and B using Lane- Eynon method.
- ❖ Value of glucose obtained is multiplied by 0.9 to get value of starch content present in flour.

**Procedure:** conical flask  $\Rightarrow$  3 g flour sample + 50 ml cold water  $\Rightarrow$  stir  $\Rightarrow$  keep it aside for 1 hr.  $\Rightarrow$  filter it and wash residue with sufficient water  $\Rightarrow$  heat the undissolved residue for 2.5 hrs. in 100 ml of 2.5% HCl solution in a flask equipped with a reflux condenser  $\Rightarrow$  Cool, neutralize with NaOH and make up volume of 250 ml and filter it  $\Rightarrow$  Fill the filtrate in burette  $\Rightarrow$  Take 5 ml each of Fehling A and B solutions in a 250 ml conical flask  $\Rightarrow$  Add 20 ml of water and few pumice stones  $\Rightarrow$  boil on burner  $\Rightarrow$  Add sugar solution from burette until a faint blue colours remain  $\Rightarrow$  Add 2-3 drops of methylene blue indicator and add sugar solution till red color precipitates of  $\text{Cu}_2\text{O}$  is produced  $\Rightarrow$  Record the volume of sugar soln used  $\Rightarrow$  reduction of Fehling solns.

- ❖ Repeat the titration using standard glucose solution.
- ❖ Calculate the sugar present in hydrolysate.
- ❖ Convert the value of glucose to starch by multiplying with 0.9.

## Observations:

1. Weight of sample taken =  $W$  gm
2. Final volume of starch hydrolysate =  $V_1$  ml
3. Volume of standard glucose solution used =  $V_2$  ml
4. Volume of starch hydrolysate used =  $V_3$  ml

## Calculations:

$$\% \text{ Glucose} = \frac{V_3 \times 2.5 \text{ mg glucose} \times V_1 \times 100}{V_2 \times 1000 \times W}$$

$$\% \text{ Starch} = \% \text{ glucose} \times 0.9$$

**THANKS**

The background features a series of overlapping, semi-transparent geometric shapes in various shades of pink and magenta. These shapes are primarily triangles and polygons that create a dynamic, layered effect on the right side of the image, while the left side remains a plain white space.